CITY OF ELKHART, INDIANA INDUSTRIAL WASTE QUESTIONNAIRE

SEC:	TION A.	GENERAL INFOR	MATION (Type or Print, Please)	
1.	Company	Name	ELKHART BRASS MFG. CO., INC.	
2.	Mailing	Address	P.O. BOX 1127	
3.	Address	of Premises _	1302 WEST BEARDSLEY AVENUE	
4.	Name and	l Title of Sig	ning Official WARREN DEFERBRACHE, CHIEF I.E./M	AINT. SUPVR.
5.	Wastewat	er discharges	to:	
	City sew	ver system	X	
	Private	septic system		
6.			harges to the City sewer system, check the types	
	of disch	arges:		**************************************
	x_ s	anitary	Wash water Rinse water	
		Cooling water	Process water Scrubber water	₩.
	0	ther		
	and not dischar	to the City s ged to the C ion A of this	ity discharges only to a private septic system ewer system, or if only sanitary sewage is ity sewer system, it is only necessary to fill questionnaire. Otherwise, complete entire	
7.	Contact	Official		
	Nam	e WARREN	DEFERBRACHE	
	Tit	1e CHIEF I	.E./MAINT. SUPVR.	
	Add	ress P.O. BO	X 1127	
	Pho	ne Number <u>29</u>	5-8330 EXT. 239 OR 250	
	me and t		tained in this questionnaire is familiar to my knowledge and belief, such information d accurate.	
			12-2-1983 Warren R. D. Feling	che
			Date Signature of Official	

Brief description of manufacturing or service activity on premise
Principal Raw Materials Used:
Catalysts, Intermediates:
Principal Product or Service (use Standard Industrial Classificat:
Manual if appropriate):
Appended to this questionnaire is a list of Standard Industric Classification (SIC) codes for industries currently or potential subject to USEPA preteatment regulations. List SIC codes feach of your processes that are subject to USEPA pretreatment regulations.

Type of	Discha	T06: .	Ratch		Continuou		Roth
For bat	ch disc	harges, I	ist types	, average	number of	batches/24	hrs.
and vol	ume (ga	llons) pe	r batch.				
Is ther	e a sch	eduled sh	utdown? _	<u>-</u>			
When? _				·			
Is prod	uction	seasonal?					
If yes,	explain	n indicat	ing months	s(s) of p	eak product	ion.	
Average	number	of emplo	yees per s	hift: .	lst; _	2nd;	3rd
Shift s	tert tir	mes:	14	ıt;	lst; 2nd;		3rd
Shifts	normally	y worked	each day	of the wee	ek:		
Shifts	-	y worked Mon	•		ek: Thu	Fri	Sat
Shifts	-	•	Tue		Thu		Sat
lst	-	Mon	Tue	Wed	Thu		Sat
lst 2nd	-	•	Tue	Wed	Thu		Sat
lst	-	Mon	Tue	Wed	Thu		Sat
lst 2nd 3rd	Sun	Mon	Tue	Wed	Thu		
lst 2nd 3rd	Sun	Mon	Tue	Wed	Thu		
lst 2nd 3rd	Sun	Mon	Tue	Wed	Thu		

SECTION D. WATER CONSUMPTION AND LOSS

Man Water Controls.	
Source	Quantity
	gallons per day
Water treatment processes in	use:
Chemical coagulation polymers, etc.	, including use of alum, ferric chloride
Lime softening	
Resin (ion exchange)	water softening
Filtration	
Chemical (chlorine or	r ozone) disinfection
Others	
List Water Consumption in Pla	ant:
Cooling Water	gallons per day
Boiler Feed	gallons per day
Process Water	gallons per day
Samitary System*	gallons per day
Contained in Product	gallons per day
Other ()	gallons per day
wenteum flan oan be set	imated at 10 gpd per employee.

4.	List average volume of discharge	e or water loss to:
	City Wastewater Sewer	gallons per day
	Septic Tank Discharge	gallons per day
	Surface Discharge	gallons perday
	Waste Hauler	gallons per day
	Evaporation	gallons per day
	Contained in Product	gallons per day
5.	Is Discharge to Sewer:	Intermittent Steady
6.	List average water usage for B-5 above:	SIC Processes itemized in Section
	Regulated SIC No. Brief Process De	
•		
		······································

SECTION E. SEWER CONNECTION AND DISCHARGE INFORMATION

1.		sewer outlets and flow: (assign sequential ch sewer starting with No. 1).	reference
	Reference No.	Descriptive Location of Sewer Connection or Discharge Point	Avg. Flow (gpd)
2.	complex show	aled drawing or dimensioned sketch of the state of the st	sbove and w location
•	for severs at monitoring a reference at	manhole, if any, and other possible samplined SIC process effluents. Indicate how City is taff can gain access to the sampling point of the description of the sampling point of	industrial nts. For
3.	measure (SPCC	sired to prepare a Spill Prevention Control and Plan per 40 CFR 112 or a RCRA Contingent eport has been prepared, attach copy. Copy report is required, but has not yet been per when it will be submitted.	cy Plan? attached.

SECTION F. PRIORITY POLLUTANT INFORMATION

1. Please indicate by placing an "X" in the appropriate box by each listed chemical whether it is Suspected to be Absent, Known to be Absent, Suspected to be Present, or Known to be Present in your manufacturing or service activity or generated as a byproduct. Some compounds are known by other names. Please refer to Appendix A for those compounds which have an asterisk(*).

1. ammonia														_
Asbestos (fibrous)		CHEMICAL COMPOUND	SUSPECTED Absent	KINTIN	SUSPECTED PRESENT	KNOWN	ITEM NO.	CHEMICAL COMPOUND	SUSPECTED	ABSENT	KNOWN	SUSPECTED	KNOWN	PRESENT
48.		ammonia					47.	chlorobenzene						
1. antimony (total) 50. chloroform* 3. arsenic (total) 51. chloromethane* 5. beryllium (total) 52. (2-chlorophenol* 1. cadium (total) 53. (2-chlorophenol* 2. cadium (total) 55. (4-chlorophenylmenylmenylmenylmenylmenylmenylmenylm	2.	asbestos (fibrous)		1		7	48.		Т	Ī	_	T	1	
	5.			Į.	i		49.		-				T	
5.								chloroform*				+		
5.		antimony (total)			1	T		chloromethane*	1			Т		
	5.	arsenic (total)		Ĺ				2-chloronaphthalene	Ι.			Ĭ	T	
S. Chromium (total)	_ 5.			İ	ļ				L	į		į.		
10. lead (total)		cadmium (total)		1	Į .			4-chlorophenylphenyl che	ਜ			ī	- 1	
1. lead (total)				<u> </u>					1_			1	1	
					<u> </u>	1				i		i	_i_	_
1.				<u> </u>	<u>i</u>	ł				-		L	1	
13.				1	!	1			1	į		t	3.	
	1			-	-	<u> </u>			 	į		-	-	
15.	13.			<u> </u>	ì	<u>i</u>			1	Ì		<u>; </u>	<u>. i</u>	
10.					<u> </u>	1			<u> </u>	_!			- 	
				 	! 	<u> </u>	02.	1,3-dichlorobeniene	 	_		<u>!</u>	-	
	10.	line (total)		·	 	<u> </u>			╄			<u> </u>	<u>!</u>	_
13.	-, -			<u> </u>	!	;			+-	_		∔	<u> </u>	
			<u> </u>	 	<u> </u>			dichiorodi luoromethane	+-	!		! 	ᆜ_	
20.				 	!	1			┿			! 	- -	_
1 1 1 1 1 1 1 1 1 1				i T		!			 			1	- -	
22. anthracene	33			-	! 	 			}	{		:	+	-
1. 1.2-dichloropropane*		بالتاقة والمستقل المستقل والمستقل والم والمستقل والمستقل والمستقل والمستقل والمستقل والمستقل والمستقل				†			7	_		<u> </u>	+	<u> </u>
21. benzidine 25. benzo(a)anthracene* 26. benzo(a)pyrene* 27. benzo(b)fluoranthene 28. benzo(b)fluoranthene 29. benzo(b)fluoranthene* 30. a-3HC (asks) 31. b-8HC (bess) 32. d-3HC (asks) 33. g-3HC (gama) 34. bis(2-chloroethoxymethane* 35. bis(2-chloroethoxymethane* 36. bis(2-chloroethoxymethane* 37. bis(2-chloromethyl)ether* 38. bis(2-chloromethyl)ether* 39. bis(2-chloromethyl)ether* 31. bis(2-chloromethyl)ether* 32. bis(2-chloromethyl)ether* 33. l,2-dipnenylhydrazine* 34. bis(2-chloromethyl)ether* 35. bis(2-chloromethyl)ether* 36. bis(2-chloromethyl)ether* 37. bis(chloromethyl)ether* 38. bis(2-chloromethyl)ether* 39. bis(andosulfan if* 39. bromodichloromethane* 39. endosulfan sulfate 39. endosulfan sulfate 39. endosulfan sulfate 39. endosulfan sulfate 39. ethylbenzene 39. fluoranthene 39. fluoranthene	77	المستقد		-	<u> </u>	┼			╄	-		÷		_
25. benzo(a)anthracene* 26. benzo(a)pyrene* 27. benzo(b)fluoranthene 28. benzo(g,h,i)perylene* 29. benzo(g,h,i)perylene* 30. a-8HC,(a,deha) 31. b-8HC (beta) 32. d-8HC (ceta) 33. d-8HC (ceta) 34. bis(2-chioroethylether* 35. bis(2-chioroethylether* 36. bis(2-chioroethylether* 37. bis(chioroethylether* 38. bis(2-chioroethylether* 39. bis(andromethylether* 31. bis(andromethylether* 32. d-8HC (ceta) 33. bis(andromethylether* 34. endosulfan II* 35. bis(andromethylether* 36. bis(andromethylether* 37. bis(andromethylether* 38. endosulfan II* 39. bromodichloromethane* 40. bromoform* 41. bromomethane* 42. 4-bromomethane* 43. endosulfan sulfate* 44. carbon tetrachloride* 45. chlordane 46. fluoranthene 47. dientyl phthalate* 48. endosulfan sulfate* 49. fluoranthene 44. carbon tetrachloride* 49. fluoranthene 45. chlordane	71			`	+ -		1 77		+-			! -		-
25. benzo(a)pyrene* 17. benzo(b)fluoranthene 17. diethyl ohthalate* 18. benzo(g,h,i)perylene* 19. benzo(k)fluoranthene* 17. diethyl ohthalate* 19. benzo(k)fluoranthene* 17. dimethyl phthalate 18. di-n-butyl ohthalate* 18. di-n-butyl ohthalate* 18. di-n-octyl ohthalate* 18. di-n-o					- -	+	1			- [ĺ
13.	76			-		Ť-	73.		┿┈	-		}	÷	_
15.	-	benzo(b)fluoranthene			1				+	1			÷	
19. benzo(k)fluoranthene* 16. dimethyl phthalate 30. a-8HC (affha) 17. di-n-butyl phthalate 31. b-8HC (bess) 18. di-n-octyl phthalate* 32. d-8HC (affsa) 19. 4.6dinitro2-methylphenol* 33. g-8HC*(gama) 80. 2,4-dinitrophenol 34. bis(2-chloroethyllether* 15. 2,4-dinitrotoluene 35. bis(2-chloroethoxymethane* 82. 2,5-dinitrotoluene 36. bis(2-chloroethoxymethane* 18. 2,4-dinitrotoluene 37. bis(chloromethyllether* 18. 38. 39. 39. 40. 41. 42. 43. 44. 45. 46. 57. 48. 49. 40. 41. 42. 43. 44. 45. 46. 47. 48. 49.	3.				T	1.			Ť	-:		:	十	_
31. b-8HC (seta) 31. b-8HC (seta) 32. d-8HC (seta) 33. g-8HC*(gema) 34. bis(2-chloroethyllether* 35. bis(2-chloroethyllether* 36. bis(2-chloroethyllether* 37. bis(chloromethyllether* 38. bis(2-chloromethyllether* 39. bis(chloromethyllether* 30. bis(2-chloromethyllether* 31. chloromethyllether* 32. chloromethyllether* 33. l,2-diphenylhydrazine* 34. endosulfan [i* 35. bis(2-ethylhexylphthaliste* 36. endosulfan [i* 37. bromodichloromethane* 37. bromodichloromethane* 38. endrin aldehyde 39. endrin aldehyde 39. chylheniene 39. fluoranthene 39. chlordane 39. heptachlor	.9.			i i	†				i	;		: 	+	_
31. b-8HC (bess) 32. d-8HC (delts) 33. g-8HC (gems) 34. bis(2-chloroethyllether) 35. bis(2-chloroethoxymethans) 36. bis(2-chloroethoxymethans) 37. bis(chloromethyllether) 38. bis(2-chloromethyllether) 39. bis(chloromethyllether) 39. bromodichloromethans 39. bromodichloromethans 30. chloromethyllether) 31. chloromethyllether) 32. chloromethyllether) 33. chloromethyllether) 34. endosulfan [[] 35. bis(2-ethylhexyllphthalists) 36. endosulfan [[] 37. bromodichloromethans 36. endosulfan sulfats 37. lendrin 38. endrin sldehyde 38. endrin sldehyde 39. chloromethans 39. fluoranthens 39. fluoranthens	30.	a-SHC (asina)		1	i	1			ì	į		1	Ì	
32. d-8HC (delta) 79. 4.6dinitro2-methylphenol* 35. g-8HC*(gamma) 80. 2.4-dinitrophenol 34. bis(2-chloroethoxymetham* 82. 2.6-dinitrotoluene 36. bis(2-chloroethoxymetham* 82. 2.6-dinitrotoluene 37. bis(2-chloroisopropylethe* 83. 1.2-diphenylhydrazine* 38. bis(2-ethylhexylphthalist* 84. endosulfan 1.5. 39. bromodichloromethane* 85. endosulfan sulfate 40. bromoform* 87. endrin 41. bromomethane* 88. endrin sldehyde 42. 4-bromophenylphenyl etazi 89. ethylheniene 43. butylbenzyl phthalate 90. fluoranthene 44. carbon tetrachloride* 91. fluorene* 45. chlordane 92. heptachlor	31.	5-8HC (besa)				ī	*8.		<u> </u>	. 1		!	1	
53. g-8HC*(gamma) 50. 2,4-dinitrophenol 54. bis(2-chloroethoxymethans 51. 2,4-dinitrotoluene 55. bis(2-chloroethoxymethans 52. 2,6-dinitrotoluene 56. bis(2-chloroisopropylethas	32.	d-SHC (40) ta)		·	1				1	1		L	j	
55. bis2-chloroethoxymethans 52. 2,5-dinitrotoluene-	33.	g-8HC*(gamma)			i			2,4-dinitrophenol	1					
35. 1,2-diphenylhydrazine* 35. 1,2-diphenylhydrazine* 35. 51s(chloromethyliether* 34. endosulfan 1 35. bis2-ethylhexylphthalste* 35. endosulfan 1 35. endosulfan 1 35. endosulfan sulfate 10. bromotorm* 87. endrin 1 11. bromomethane* 88. endrin sldehyde 12. 4-bromophenylphenyl ethan 15. ethylhenzene 15. butylbenzyl phthalate 190. fluoranthene 14. carbon tetrachloride* 191. fluorene* 15. chlordane 192. heptachlor		bis(2-chloroethyDether		T					I			Ĺ		
57. bis(chloromethyliether" 54. endosultan [" 58. bis2-ethylhexylphthalste" 55. endosulfan [[" 59. bromodichloromethane" 56. endosulfan sulfate 10. bromoform" 87. endrin 11. bromomethane" 58. endrin aldehyde 12. 4-bromophenylphenyl etmm 59. ethylheniene 13. butylbenzyl phthalate 90. fluoranthene 14. carbon tetrachloride" 91. fluorene" 15. chlordane 92. heptachlor				j					1					
35. bis2-ethylhexylponthalster 35. endosulfan []* 39. bromodichloromethane* 36. endosulfan sulfate 10. bromoform* 87. endrin 11. bromomethane* 88. endrin aldehyde 12. 4-bromophenylphenyl ettat 59. ethylheniene 13. butylbenzyl phthalate 90. fluoranthene 14. carbon tetrachloride* 91. fluorene* 15. chlordane 92. heptachlor	36.	bist-chlordsopropylather												
39. bromodichloromethane*	37.	bis (chloromethyl)ether			-					_			\Box	
10. bromoform* 87. lendrin 11. bromomethane* 88. lendrin aldehyde 12. 4-bromomethane* 89. lethylheniene 13. butylbenzvi phthalate 90. fluoranthene 14. carbon tetrachloride* 91. fluorene* 15. chlordane 92. heptachlor				<u> </u>	1	<u> </u>			!	<u></u>		<u>i </u>	<u> </u>	
11. bromomethane* 12. 4-bromophenylphenyl etta: 55. endrin aldehyde 13. butylbenzyl phthalate 90. fluoranthene 14. carbon tetrachloride* 91. fluorene* 15. chlordane 92. heptachlor				ــــــــــــــــــــــــــــــــــــــ	1	<u> </u>	: 36,		;	_1		i		
42. 4-bromophenylphenyl edsz			<u> </u>	<u> </u>	<u> </u>	<u> </u>			i	1				
13. butylbenzvi phthalate 90. fluoranthene 14. carbon tetrachloride 91. fluorene 15. (chlordane 92. heptachlor				1	 					į		<u> </u>	-	
44. carbon tetrachloride" 191. fluorene" 15. chlordane 192. heptachlor				1	<u> </u>	 	-		1	_		<u> </u>		
45. (chlordane 92. heptachlor				 	!				٠	_		ļ	1	
				!	<u>!</u>					_		1		
46. H-ChigTO-J-RethylDRengi7				1	1	!								_
	46.	H-chioro-3-methylphenol	· ·	1	1		93.	heptachlor epoxide		_		i		

SECTION F. PRIGRITY POLLUTANT INFORMATION (CON'T)

ITEM NO.	CHEMICAL COMPOUND	SHISPECTED ABSENT KNOWN	SUSPECTED	KNOWN	ITEM NO.	CHEMICAL COMPOUND	SUSPECTED ABSENT	KWOWN	SUSPECTED	KNOWN
94.	hexachlorobenzene*	<u> </u>		_[]	112.	PCB-1248*]	
95.	hexachlorobutadiene				113.	PCB-1254*	<u></u>			1
96.	hexachlorocyclopenta-		1		114.	PCB-1260*	Ī			T
	diene*				115.	pentachlorophenol	i			1
9	hexachloroethane"				116.	phenanthrene	I			
98.	indeno(1,2,3-cd)pyrene*				117.	phenol	T			T
99.	isophorone"				118.	pyrene				
100.	methylene chloride				119.	2,5,7,8-retrachlorodi-				
	naphthalene					benzo-p-dioxin*	Ţ.			
102.	initrobenzene				120.	1122-tetrachloroethane	j			
105.	: 2-nitrophenol*				121.	tetrachloroethene*				T
104.					122.	toluene"	Ĭ		1	
	n-nitrosodimethylamine*				123.	toxaphene				<u> </u>
106.	n-nitrosodipropylamine*				124.	1,2,4-trichlorobenzene			<u> </u>	<u> </u>
107.				<u> </u>	125.	LLitrichloroethane"			<u> </u>	
108.				11	126.	l, l, 2-trichloroethane*	1		-	
109.		1	<u>.</u>	4	127.	trichloroethene*	<u> </u>			<u> </u>
110.					123.	trichlorofluoromethan@	<u> </u>		<u> </u>	
111.	1 PCB-1242*			4	139.	2,4,6-trichlorophenol				<u> </u>
	1	l î	1	1 1	130.	vinyl chloride*	•		i —	1

2. For chemical compounds in F-2 above which are indicated to be "Known Present," please list and provide the following data for each: (attach additional sheets if needed).

ITEM NO.	CHEMICAL COMPOUND	ANNIAL. HSAGE (LBS)	ESTIMATED LOSS TO SEMEN LBS. /YR.	ITEM NO.	CHEMICAL COMPOUND	AVBRIAL PISACE (LBS)	ESTINATED LOSS TO SEMER LBS./YR.
	· · · · · · · · · · · · · · · · · · ·			<u>:</u>		-	ļ <u>.</u>
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			1				
						 	
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4.	Describe, what if any, laboratory analyses have been conducted on process waste streams in the plant, including which streams were sampled, what parameters were measured, and frequency and type of samples. (The baseline report referred to in G2 below can be referenced in answering this question.)
SEC	TION C DEPTERATMENT
1.	TION G. PRETREATMENT Is this plant subject to an existing Pretreatment Standard?
1.	
	Is this plant subject to an existing Pretreatment Standard? Is this plant required to submit a baseline report per 40 CFR 403.12? If a baseline report has been prepared, attach a copy to this questionnaire. Copy attached If a baseline
2.	Is this plant subject to an existing Pretreatment Standard? Is this plant required to submit a baseline report per 40 CFR 403.12? If a baseline report has been prepared, attach a copy to this questionnaire. Copy attached. If a baseline report is required, but has not yet been prepared, indicate date that it will be submitted. If subject to Federal Pretreatment Standards, are the standards
	Is this plant subject to an existing Pretreatment Standard? Is this plant required to submit a baseline report per 40 CFR 403.12? If a baseline report has been prepared, attach a copy to this questionnaire. Copy attached. If a baseline report is required, but has not yet been prepared, indicate date that it will be submitted. If subject to Federal Pretreatment Standards, are the standards being met on a consistent basis? (The baseline report can be

4.	Are additional pretreatment facilities and/or operation and maintenance required to meet Pretreatment Standards? If additional pretreatment							
	and/or operation and maintenance are required, list the schedule							
	by which they will be provided. (The baseline report can be referred							
	to in answering this question.)							
5.	Describe residuals (sludges, precipitates, etc.) that are produced							
	or result at your facility and the methods employed to dispose of the residuals. List names of waste haulers, if applicable.							
	of the restrogis. That names of waste nagrets, it applicants,							
	TRAMP OIL AND DEGREASER SLUDGE							
	GOLD SHIELD SOLVENTS							
	2263 DISTRIBUTORS DRIVE							
	INDIANAPOLIS, IN 46241							

E. S. A. INDUSTRIAL WASTE SURVEY PRELIMINARY ANALYSIS SHEET

Date 8-16-83
Sampling Point #1 ELKHART BRASS.

P.O. BOX 1127
Cekhart, Ind. 40515

monia Nitrogen.....

No. of Samples

Type of Sample <u>Composite</u>

Hrs. of Sampling <u>24</u>

No. of Sample Points

	consast, no	· 703/3	
Parameters	Analyze For	Parameters	Analyse Per
рн	P, 4	Arsenic	
	41	Cadmium	*.10
COD	•••••	Chromium hexavales	14
Chlorine Demand	••••••	Chromium total	
Color	••••••	Copper	
Total Solids	······································	Lead	*.!0
TS8	26.	Iron	
	•••••	Manganese	The second of th
Grease & Gil.u	•••••	Mercury	
Phenols	• • • • • • • • • • • • • • • • • • • •	Nickle	10 mm 40 20 64 20 13
Chloride		Zinc	
Sulfate			
Sulfides			
Total Phosphorus	,30		

E. S. A. INDUSTRIAL WASTE SHRVEY PRELIMINARY ANALYSIS SHEET

Date 8-17-83
Sampling Point #1 ELKHART BRASS

No. of Samples
Type of Sample Composite
Hrs. of Sampling 2.7
No. of Sample Points

DAVADATA	Analusa Eng	Damanadana	
Parameters	Analyze For	Parameters	Anal
рн	7,7	Arsenic	• • • • • • • • • •
BOD (5 day)	/2	Cadmium	
COD		Chromium hexav	lent
Chlorine Demand		Chromium total	
, Color		Copper	
Total Solids	··	Lead	
788	32	Iron	
Settleable Solids		Manganese	
Grease & Cil.u		Mercury	
Phenols	••••••	Nickle	
Chloride		Zinc	
Sulfate			
Sulfides	••••••		
Total Phospherus	.50		
Cyanide			
T.O. C.	· · · · · · · · · · · · · · · · · · ·		

R. D. INGRAHAM, Manager



WASTEWATER TREATMENT PLANT 1201 South Nappanee Street Elkhart, Indiana 46516 (219) 293-2572

May 9, 1986

Mr. Warren DeFerbrache Elkhart Brass Mfg. Co., Inc. P.O. Box 1127 Elkhart, Ind. 46515

Dear Mr. DeFerbrache:

This is in response to your verbal request to Mr. Larry Pozgay to discharge water from two underground tanks into the City sewer system.

Based on the test results of this water which you submitted, the Wastewater Treatment Plant would be happy to accept the discharge from the 20,000 gallon tank which contained no Volatile Organic Compounds at detectable levels.

We are unable, however, to accept the discharge at this time from the 12,000 gallon tank due to the presence of various toxic substances. We would suggest that Elkhart Brass contract a licensed hazardous waste hauler to remove this waste. This would protect the Wastewater Treatment Plant from receiving any hazardous waste and would provide Elkhart Brass with certification that the waste was disposed of in the proper manner.

We would like to thank Elkhart Brass for their co-operation with the City in this matter. Please contact either Mr. Pozgay or me should you have any questions or comments. Thank you.

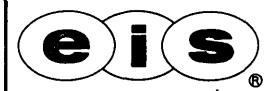
Sincerely,

David A. Bates

Pretreatment Director

COPIES:

R. D. Ingraham Larry Pozgay -



EIS ENVIRONMENTAL ENGINEERS, INC.

1701 North Ironwood Drive . South Bend, Indiana 46635 . Telephone (219) 277-5715

VOLATILE ORGANIC COMPOUND (VOC) ANALYSIS REPORT

5-1-86 Client: Elkhart Brass Date Reported: EIS Lab No.: 1639F, 1640F 4-29-86 c/o Warren R. Deferbrache Sample Date: 4-29-86 Date Received: P.O. # 04463 Date Analyzed: 4-30-86 Samples Received Sample ID: Project Number 1268-02 Refrigerated: Yes No

In 40cc Vials: Yes Air Space: Underground Tank #1 20000 gallons Yes

12000 gallon Tank #2

RESULTS

- The test procedures used for this analysis, and the listing of compounds detectable by these procedures, are described in Table 1 on the reverse side of this report sheet.
- If your sample contained any of the Table 1 Priority Pollutant Volatile Organic Compounds above a Quantifiable Detection Limit of µg/l, these compounds are reported below. If no Table 1 Volatile Organic Compounds were detected, then a statement to this effect is listed below.
- Results are as follows:

20000 Gallon Underground Tank

No Volatile Organic Compounds were detected at a Quantifiable Detection Limit of 10 μ g/1.

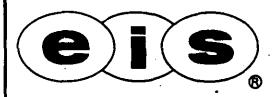
The Ignitability Test resulted in No Flash Point but the sample itself ignited at 139 $^{\circ}\text{F}$.

12000 Gallon Tank

Parameter	Concentration	$(\mu g/1)$
Ethyl Benzene	95	
Toluene	170	
Xylenes .	570	
Multicomponent Residue*	20500	
1,1,1-Trichloroethane	180	
	•	•

Defined as a "species" similar to fuel oil or paint residuals and quantitated using a Toluene standard.

This sample has a Flash Point >162 °F.



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This sample has a Flash Point >162 *F.

Toluene- haz. substance twaste

Ethyl Benzene-haz. subst.

+ priority Toxic pollute

ABORATORY DIRECTOR

1,1,1- Has vaste-private toxic pollutant

TABLE 1

PARTIAL LISTING - VOLATILE ORGANIC COMPOUNDS SPECIES DETECTABLE USING METHODS STATED BELOW

-- PRIORITY POLLUTANTS -----

- NON PRIORITY POLLUTANTS -

Benzene
Bromodichloromethane
Bromoform
Bromomethane
Carbon Tetrachloride
Chlorobenzene
Chloroethane
2-Chloroethylvinyl Ether
Chloroform
Chloromethane
Dibromochloromethane
1,2-Dichlorobenzene
1,3-Dichlorobenzene
1,4-Dichlorobenzene
1,1-Dichloroethane

1,2-Dichloroethane
1,1-Dichloroethylene
t-1,2-Dichloroethylene
1,2-Dichloropropane
c-1,2-Dichloropropene
t-1,2-Dichloropropene
Ethyl Benzene
Methylene Chloride
1,1,2,2-Tetrachloroethane
Tetrachloroethylene
1,1,1-Trichloroethane
Toluene
Trichloroethylene
Vinyl Chloride

Acetone
Dichlorodifluoromethane
Methyl Ethyl Ketone
Methyl Isobutyl Ketone
Styrene
Trichlrofluoromethane
Vinyl Acetate
Xylene
2-Hexanone

REFERENCES

- . "Test Methods: Methods for Organic Chemical Analysis of Municipal and Industrial Waste-water" USEPA-600/4-82-057, July 1982, Method 601 and Method 602
- . "Test Methods for Evaluating Solid Waste, Physical/Chemical Methods" SW-846, July 1982, Methods 5030, 8010, 8020.

ANALYTICAL PROCEDURES

- . Purge and Trap, Gas Chromatography is utilized.
- . The effluent from the gas chromatographic column is monitored by Photoionization and Hall 700A Electrolytic Conductivity Detectors operating in series.
- . Surrogate compounds are added prior to the Purge step to monitor overall system performance. The surrogates also function as Retention Time Standards.
- . Quantitation is made by external standards.
- . Identification is made by relative retention times and responses to the two in series detectors.

R. D. INGRAHAM, Manager



WASTEWATER TREATMENT PLANT 1201 South Nappanee Street Elkhart, Indiana 48516 (219) 293-2572

June 6, 1986

Mr. Warren DeFerbrache Elkhart Brass Mfg. Co., Inc. P.O. Box 1127 Elkhart, Indiana 46515

Dear Mr. DeFerbrache:

We would like to thank you for your time and consideration in allowing us the opportunity to inspect your facility on this date. We feel that it is very important for the City of Elkhart to work in co-operation with industry to insure the protection of our environment.

The City of Elkhart Pretreatment Staff would like to take this opportunity to thank you and your company for your co-operation and assistance with our Pretreatment Program. Please feel free to contact us with any comments or questions you may have.

Sincerely,

David A. Bates

Pretreatment Director

Larry Pozgay Laboratory Manager

COPY:

R.D. Ingraham



INDUSTRIAL INSPECTION FORM

		•		
INDUSTRY_	Elkhart Brass	Mfg. Co., Inc.	DATE June 6, 1986	 -
ADDRESS	1302 West Bear	isley Ave., Elkhart, In	nd	
TIME STAP	TED 1:30 PM	TIME	COMPLETED 2:00 P	x
PURPOSE C	F INSPECTION	Check process water usa	ige for possible SI	U
sta	tus.		·	
SAMPLE (S)	TAKENYES_	NO X TYPE		
SAMPLE TI	ME (GRAB):			
#1	······································	#6	#11	•
#2		# 7	#12	
#3		#8	#13	•
#4	 	#9	#14	*
#5		#10	#15	•
COMPOSITE	SAMPLE FROM	TO		
NOTES:	3:30 PM and se has only about Largest water ment, up to 10 contact coolin Foundry has no All undergroun site and the a Vapor degrease is located on tained trench	water usage. d storage tanks have be reas have been concrete r uses 1,1,1 Trichloro floor area. Old unit wo which has sump pump.	o 11 PM. Second shiry works only 1st sfire fighting equipment of the fire fighting equipment of the control of the fire fighting equipment of self-control of	shift. sip- on- ton-
	It would not a	ppear that this facilignates at this	ty would qualify as	

INSPECTION AND/OR SAMPLING BY Dave Bates/Larry Pozqay